

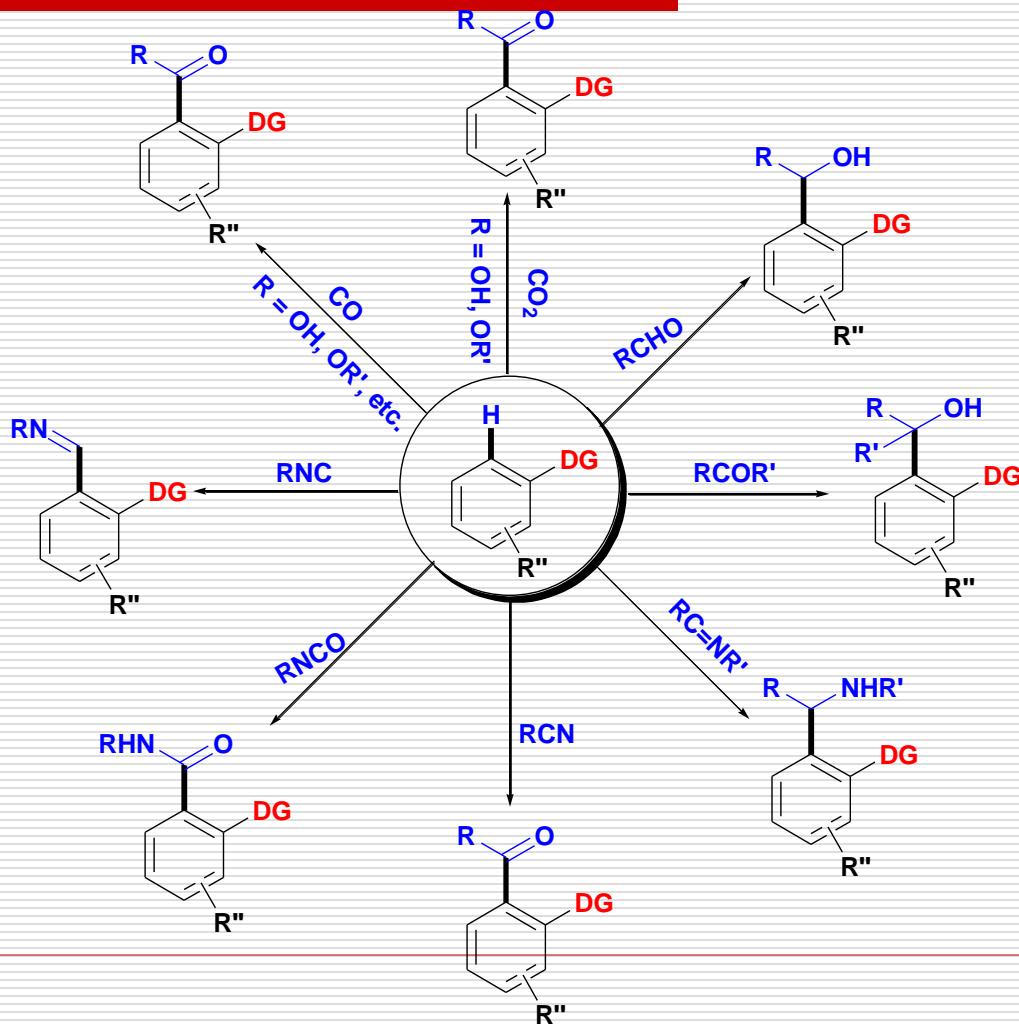
Transition-Metal-Catalyzed Additions of C-H bond to C-X (X = N, O) Multiple Bonds *via* C-H Bond Activation

Dong group, Literature report

Guobing Yan

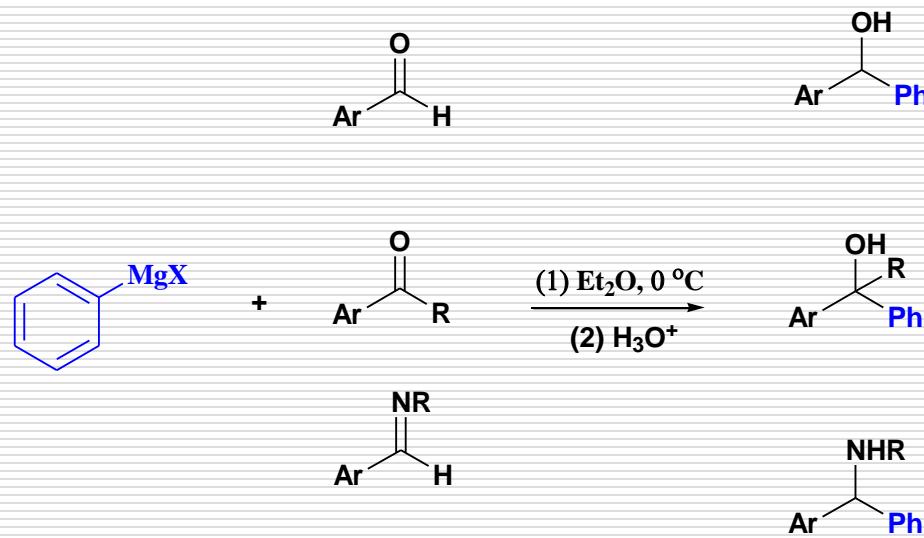
5/15/2013

Contents



1 Background

1. Nucleophilic addition of organometallic reagents to carbonyl compounds and their derivatives

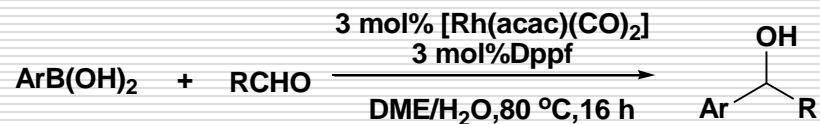


Drawbacks:

- (1) strict anhydrous conditions
- (2) imperfect functional group tolerance
- (3) prefunctionalization of nucleophilic coupling partners
- (4) unwanted formation of stoichiometric salt waste

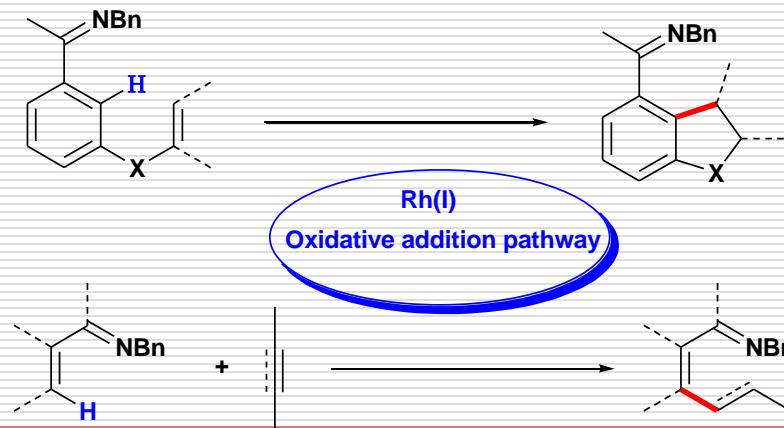
1 Background

2.Rhodium-catalyzed 1,2-addition reactions of organoboronic acids to aldehydes



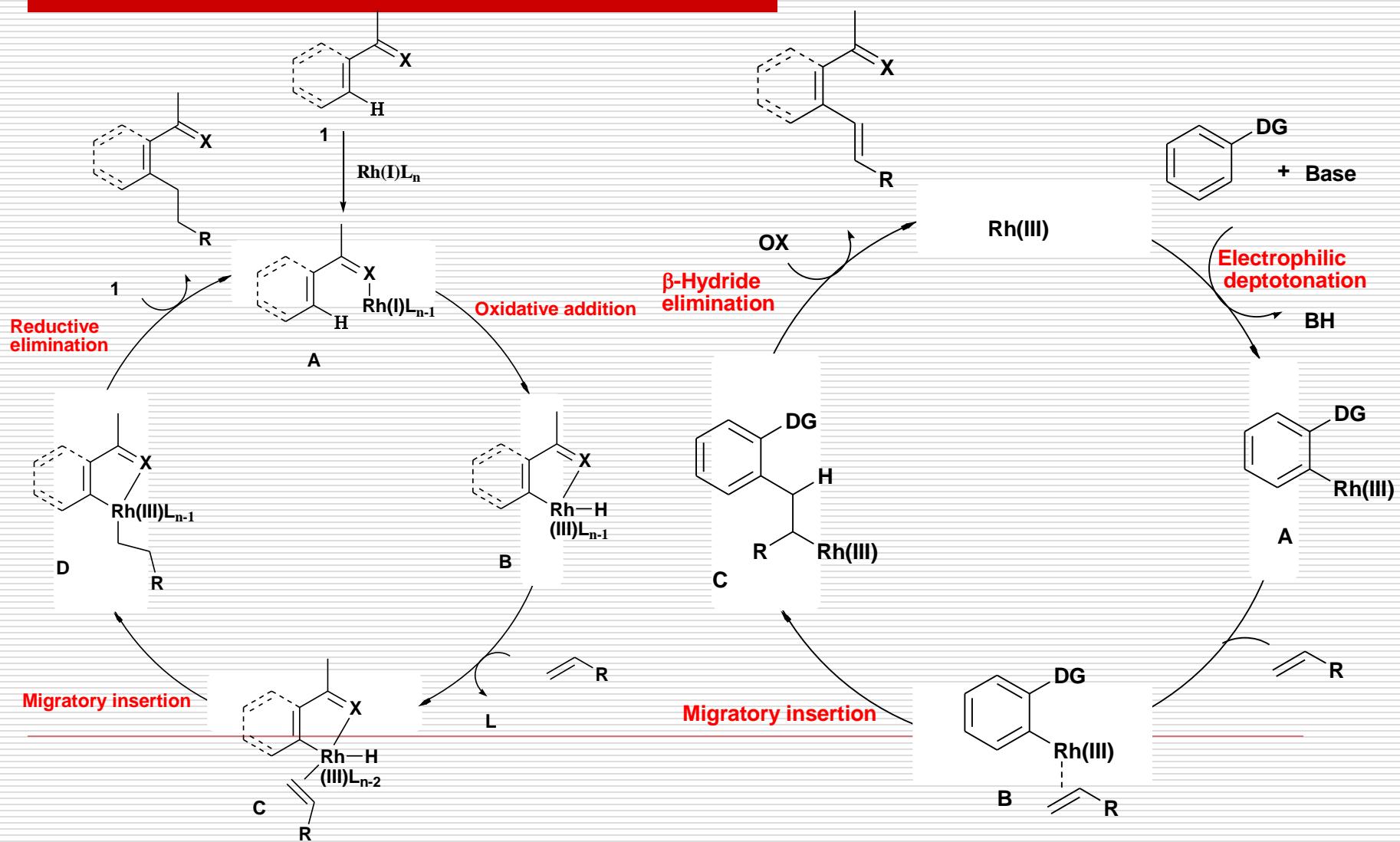
M. Sakai, M. Ueda, N. Miyaura, *Angew. Chem., Int. Ed.* 1998, **37**, 3279-3281.

3.Transition metal-catalyzed addition reactions of C-H bond to alkene and alkyne

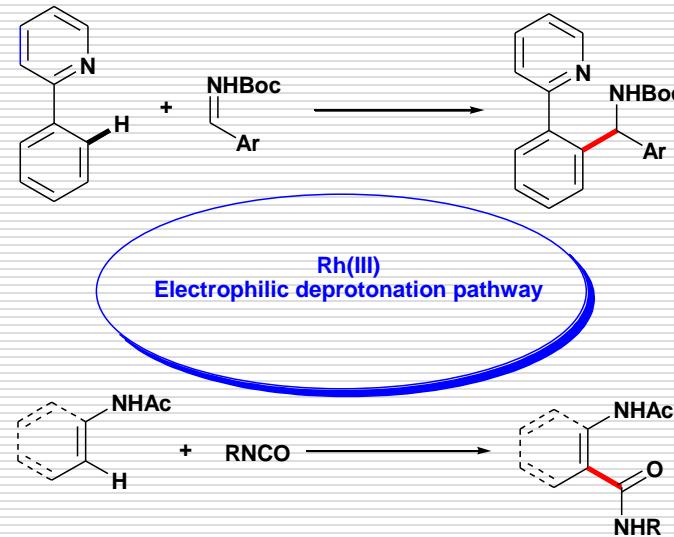


D. A. Colby, A. S. Tsai, R. G. Bergman, J. A. Ellman, *Acc. Chem. Res.* 2012, **45**, 814-825

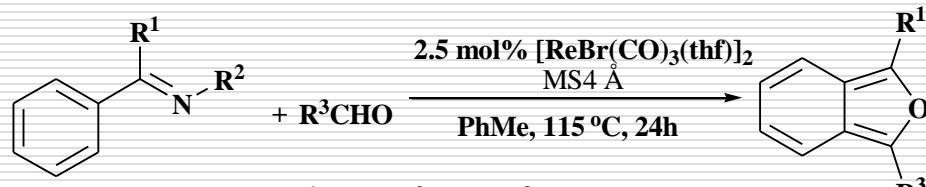
1 Background



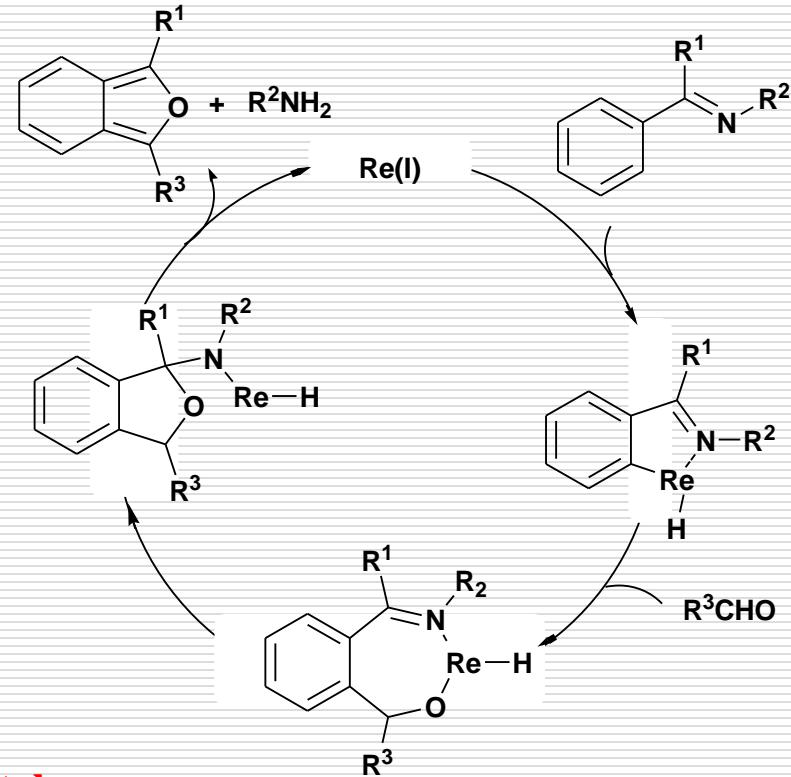
1 Background



2 Addition of C-H bonds to aldehydes



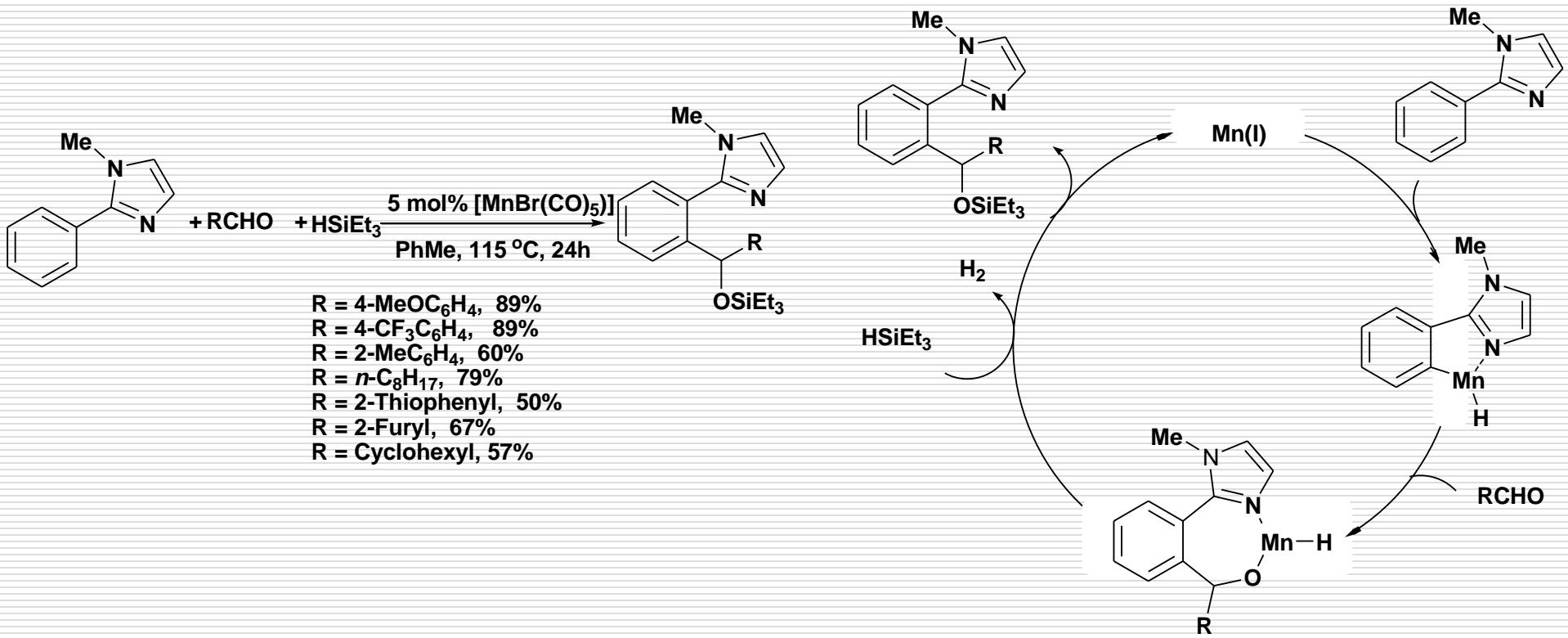
$\text{R}^1 = \text{Ph}, \text{R}^2 = \text{Ph}, \text{R}^3 = \text{Ph}, 95\%$
 $\text{R}^1 = \text{Ph}, \text{R}^2 = \text{Bn}, \text{R}^3 = \text{Ph}, 95\%$
 $\text{R}^1 = \text{Ph}, \text{R}^2 = \text{Ph}, \text{R}^3 = 4\text{-MeOC}_6\text{H}_4, 76\%$
 $\text{R}^1 = \text{Ph}, \text{R}^2 = \text{Ph}, \text{R}^3 = 4\text{-MeC}_6\text{H}_4, 94\%$
 $\text{R}^1 = \text{Ph}, \text{R}^2 = \text{Ph}, \text{R}^3 = 4\text{-CF}_3\text{C}_6\text{H}_4, 98\%$
 $\text{R}^1 = \text{Ph}, \text{R}^2 = \text{Ph}, \text{R}^3 = 2\text{-MeC}_6\text{H}_4, 90\%$



Rhenium has some properties of early and late transition metals

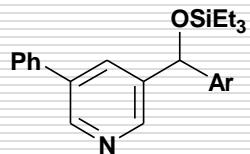
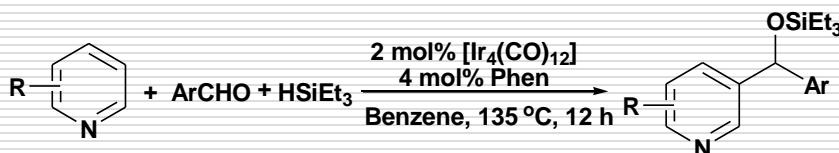
Notes: (1) carbon-rhenium bonds show nucleophilicity of early transition metals
(2) the rhenium complex can also take place reductive elimination of late transition metals

2 Addition of C-H bonds to aldehydes

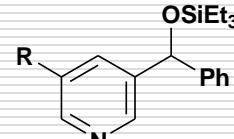


Notes: (1) polarity of the manganese–carbon bond
(2) asymmetric transformation using an aromatic compound with a chiral substituent

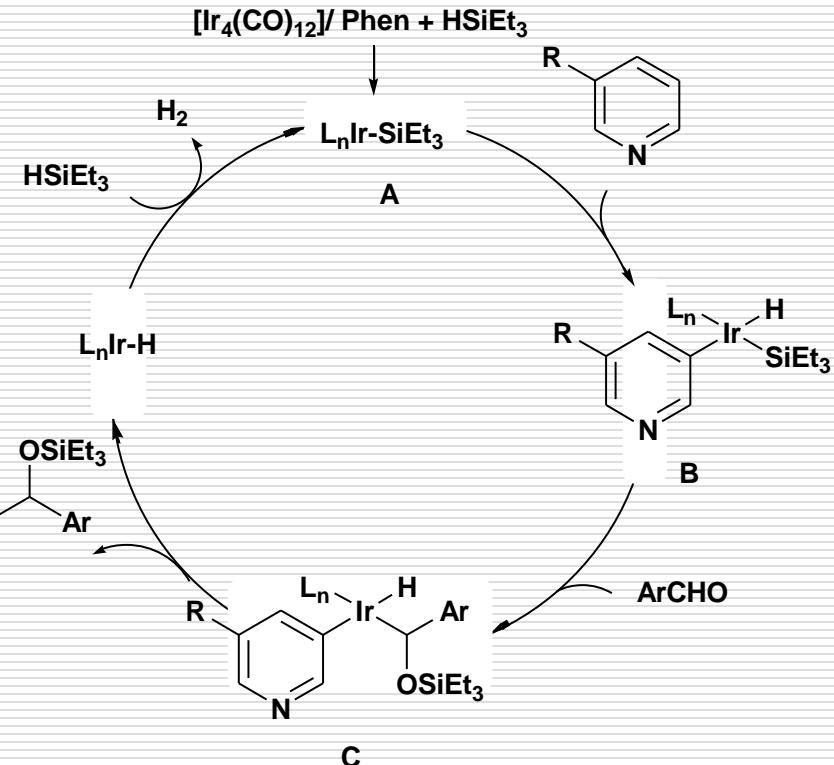
2 Addition of C-H bonds to aldehydes



Ar = C₆H₅, 73%
 Ar = 2-MeC₆H₄, 78%
 Ar = 4-MeOC₆H₄, 57%
 Ar = 4-FC₆H₄, 74%
 Ar = 4-ClC₆H₄, 62%
 Ar = 2-Naphthyl, 72%
 Ar = 2-Benzofuryl, 55%

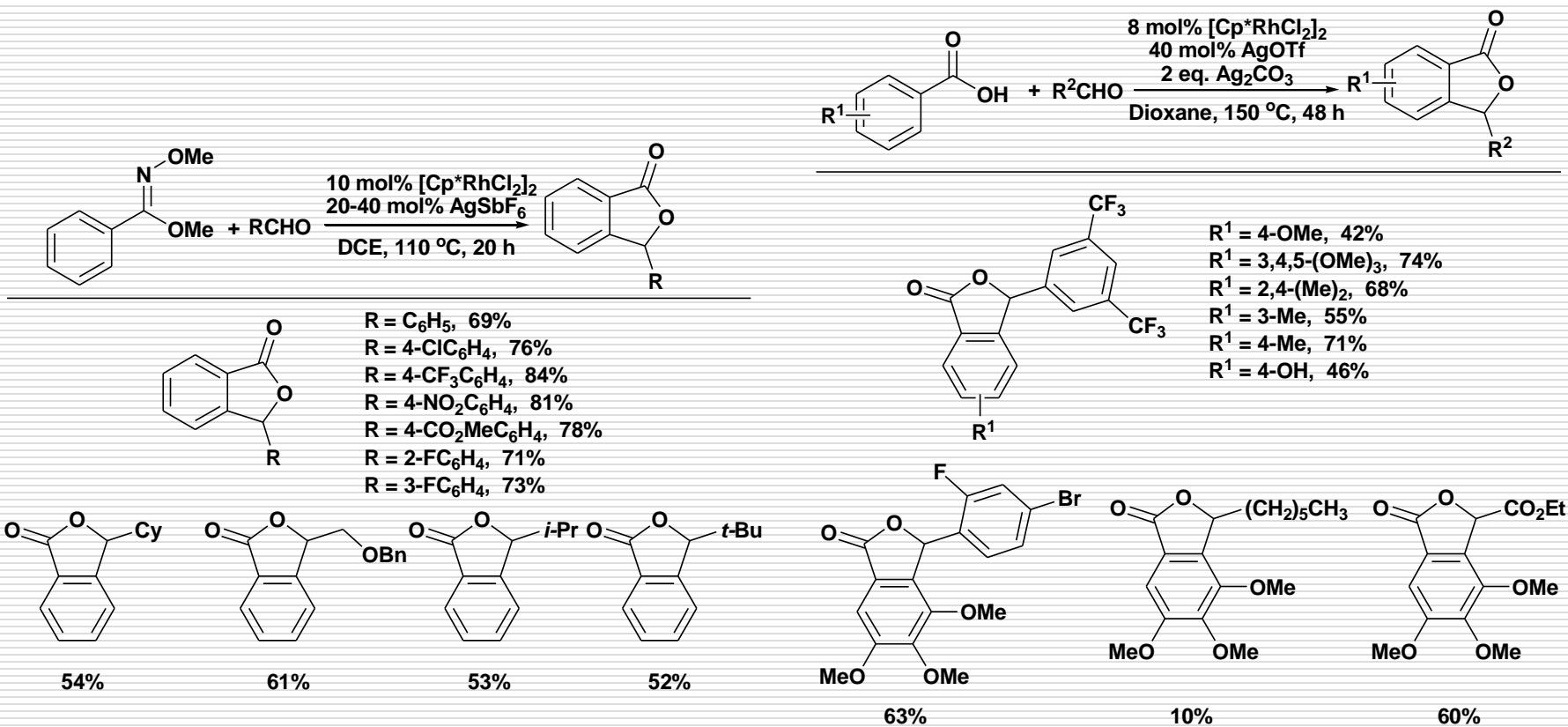


R = 4-FC₆H₄, 68%
 R = 4-MeOC₆H₄, 66%
 R = Me, 61%
 R = Bn, 66%
 R = OMe, 53%
 R = OBn, 55%



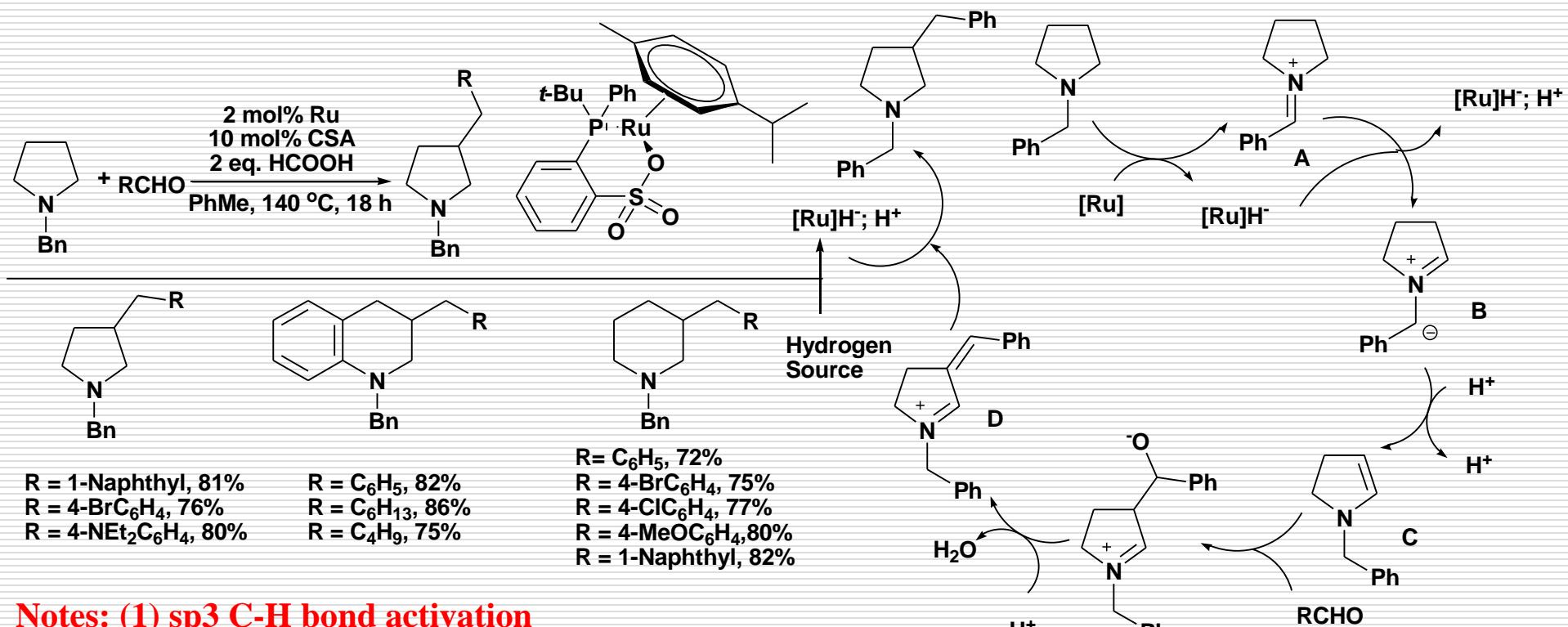
- Notes:
- (1) trialkylsilane was essential for the reaction
 - (2) a silyl iridium complex might be the active catalytic intermediate
 - (3) unusual meta selectivity
 - (4) without directing group

2 Addition of C-H bonds to aldehydes



Notes: Synthesize biologically important phthalides in a single step.

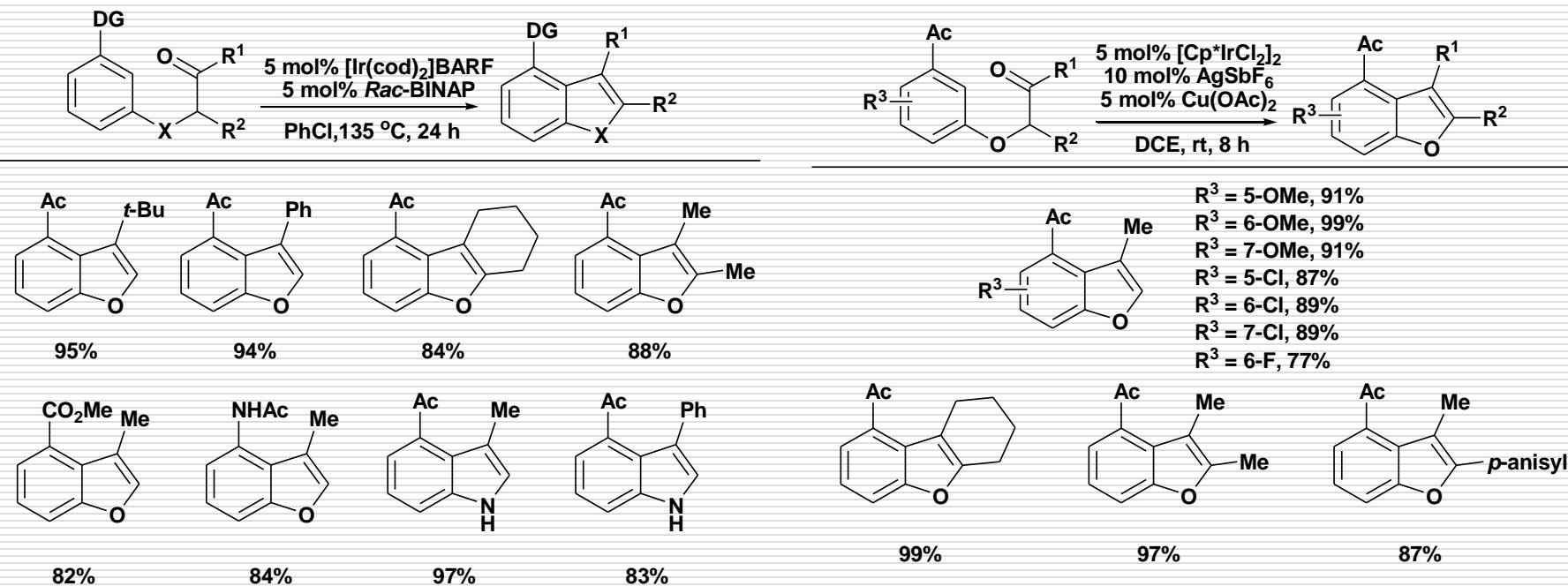
2 Addition of C-H bonds to aldehydes



Notes:

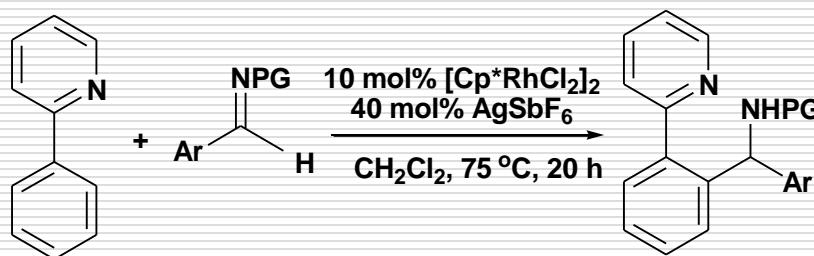
- (1) sp³ C-H bond activation
- (2) highly regioselective transformation
- (3) camphor sulfonic acid was essential for the reaction

3 Addition of C-H bonds to ketones

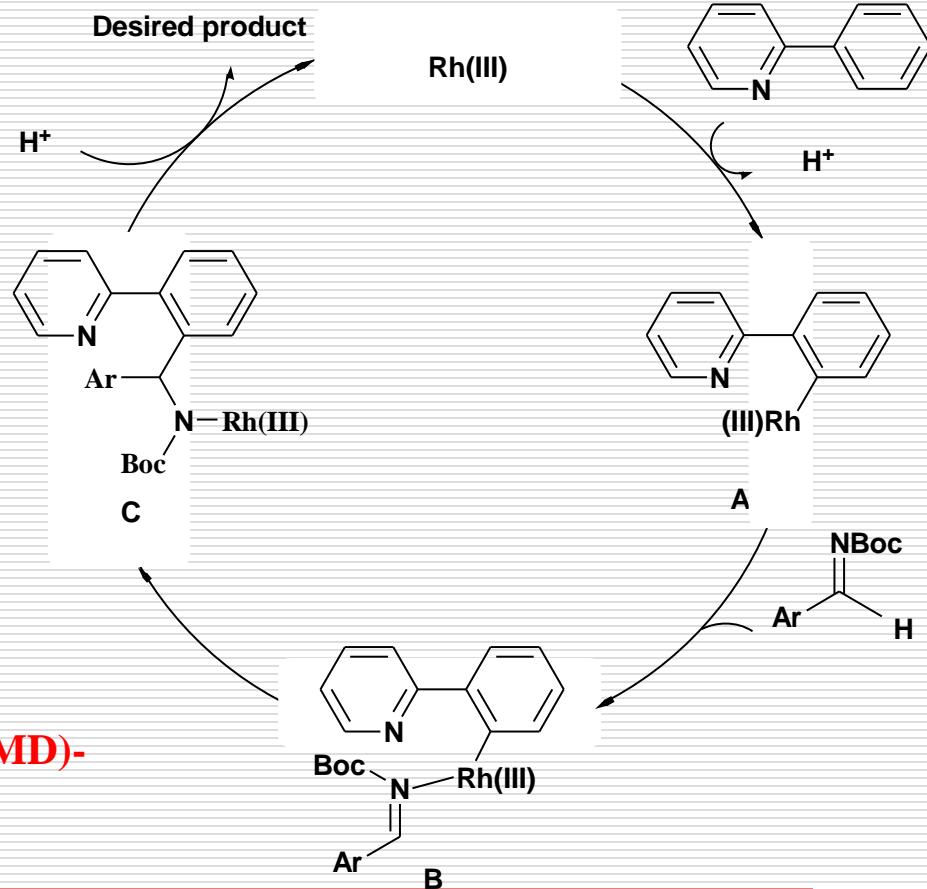


Notes: (1) Ketones are less reactive than aldehydes in this type addition reactions
(2) directed C-H bond cleavage or electrophilic metalation

4 Addition of C-H bonds to imines

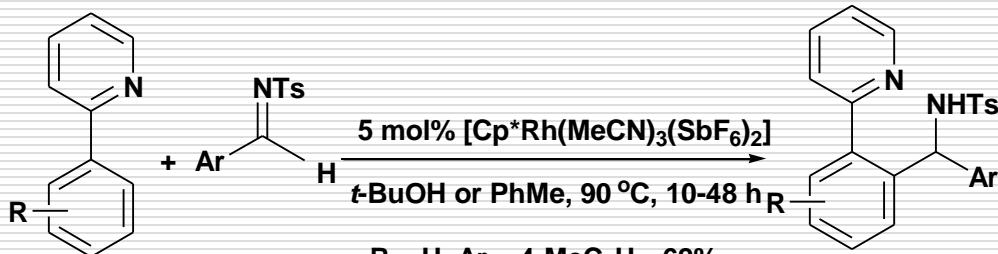


$\text{Ar} = \text{C}_6\text{H}_5, \text{PG} = \text{Boc}, 82\%$
 $\text{Ar} = 4\text{-ClC}_6\text{H}_4, \text{PG} = \text{Boc}, 77\%$
 $\text{Ar} = 4\text{-NO}_2\text{C}_6\text{H}_4, \text{PG} = \text{Boc}, 77\%$
 $\text{Ar} = 4\text{-CNC}_6\text{H}_4, \text{PG} = \text{Boc}, 50\%$
 $\text{Ar} = 4\text{-CF}_3\text{C}_6\text{H}_4, \text{PG} = \text{Boc}, 95\%$
 $\text{Ar} = 2\text{-MeC}_6\text{H}_4, \text{PG} = \text{Boc}, 92\%$
 $\text{Ar} = \text{C}_6\text{H}_5, \text{PG} = \text{Ts}, 40\%$
 $\text{Ar} = \text{C}_6\text{H}_5, \text{PG} = \text{Ns}, 51\%$

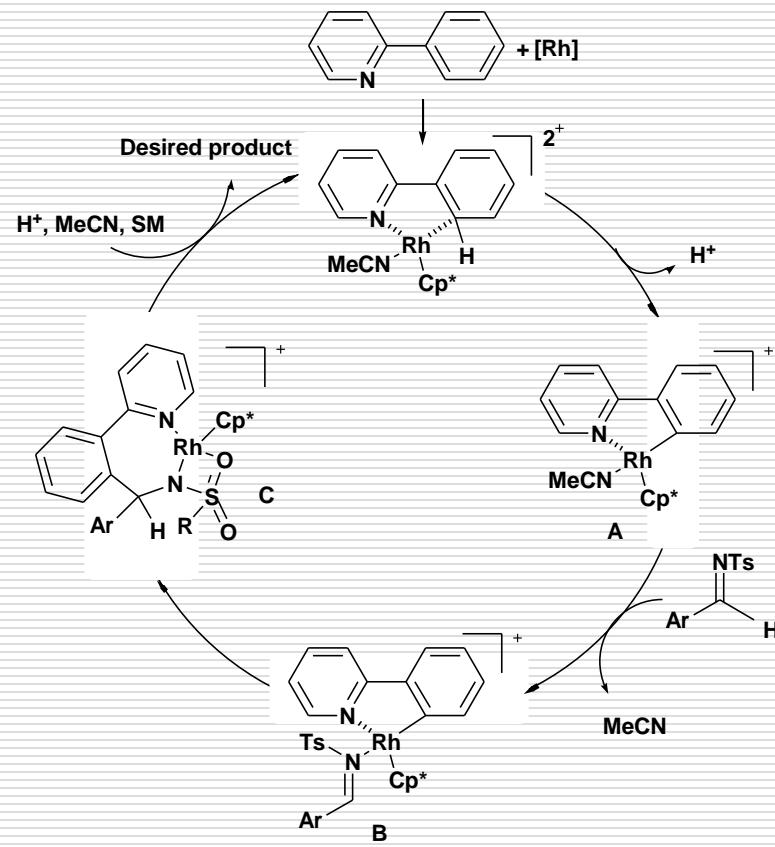


Notes: (1) **electrophilic deprotonation**
(2) **concerted metalation-deprotonation (CMD)-**

4 Addition of C-H bonds to imines

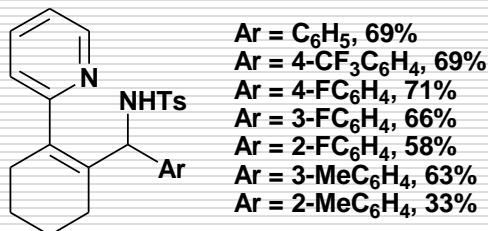
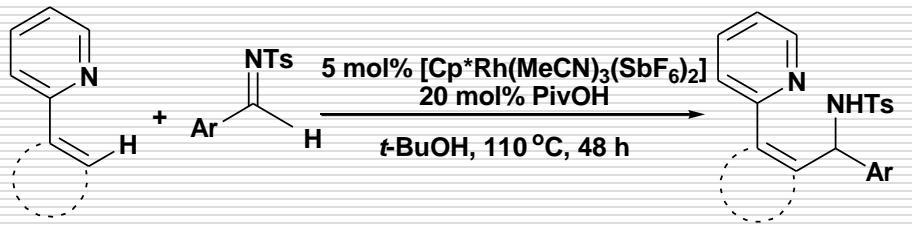


$\text{R} = \text{H}, \text{Ar} = 4\text{-MeC}_6\text{H}_4, 62\%$
 $\text{R} = \text{H}, \text{Ar} = 4\text{-BrC}_6\text{H}_4, 76\%$
 $\text{R} = \text{H}, \text{Ar} = 4\text{-ClC}_6\text{H}_4, 83\%$
 $\text{R} = \text{H}, \text{Ar} = 4\text{-FC}_6\text{H}_4, 83\%$
 $\text{R} = \text{H}, \text{Ar} = 3\text{-FC}_6\text{H}_4, 82\%$
 $\text{R} = \text{H}, \text{Ar} = 2\text{-Thiophenyl}, 82\%$
 $\text{R} = 4\text{-OMe}, \text{Ar} = \text{C}_6\text{H}_5, 81\%$
 $\text{R} = 4\text{-OH}, \text{Ar} = \text{C}_6\text{H}_5, 86\%$
 $\text{R} = 4\text{-Cl}, \text{Ar} = \text{C}_6\text{H}_5, 73\%$
 $\text{R} = 4\text{-Br}, \text{Ar} = \text{C}_6\text{H}_5, 74\%$
 $\text{R} = 4\text{-OCONMe}_2, \text{Ar} = \text{C}_6\text{H}_5, 70\%$
 $\text{R} = 4\text{-CO}_2\text{Me}, \text{Ar} = \text{C}_6\text{H}_5, 78\%$

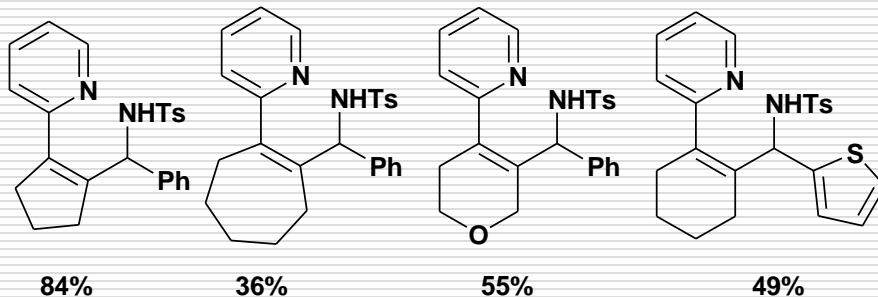


Notes: mild conditions and without any additives

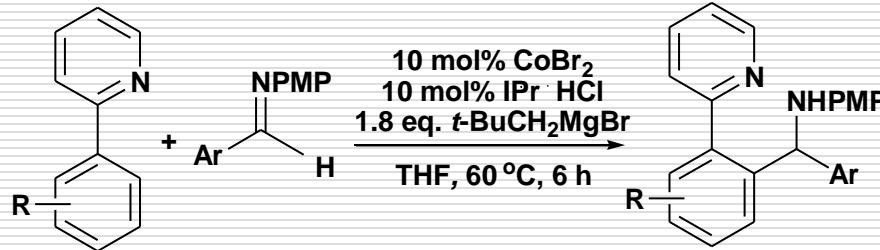
4 Addition of C-H bonds to imines



The first example of direct alkenyl C-H addition to imines and aldehydes.

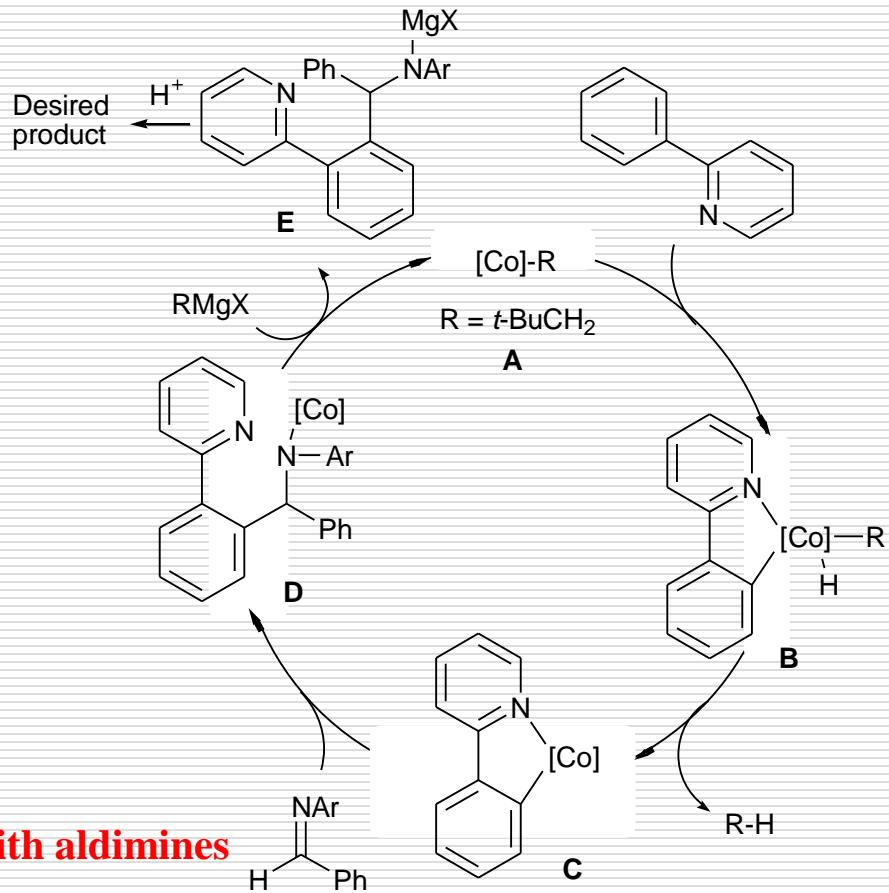


4 Addition of C-H bonds to imines



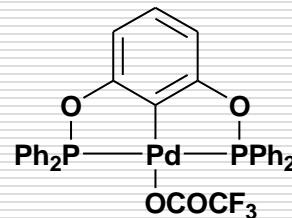
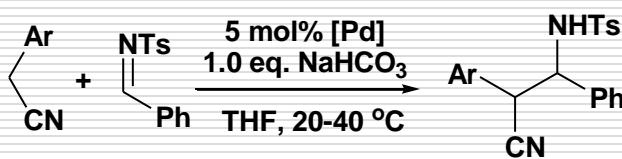
$\text{R} = \text{H}, \text{Ar} = 4\text{-MeC}_6\text{H}_4, 85\%$
 $\text{R} = \text{H}, \text{Ar} = 4\text{-MeOC}_6\text{H}_4, 78\%$
 $\text{R} = \text{H}, \text{Ar} = 4\text{-ClC}_6\text{H}_4, 83\%$
 $\text{R} = \text{H}, \text{Ar} = 4\text{-FC}_6\text{H}_4, 26\%$
 $\text{R} = \text{H}, \text{Ar} = 3\text{-MeC}_6\text{H}_4, 77\%$
 $\text{R} = \text{H}, \text{Ar} = 2\text{-MeC}_6\text{H}_4, 77\%$
 $\text{R} = \text{H}, \text{Ar} = 2\text{-MeOC}_6\text{H}_4, 53\%$

$\text{R} = \text{H}, \text{Ar} = 2\text{-C}_6\text{H}_5\text{C}_6\text{H}_4, 41\%$
 $\text{R} = \text{H}, \text{Ar} = \text{C}_6\text{H}_5, 81\%$
 $\text{R} = 4\text{-OMe}, \text{Ar} = \text{C}_6\text{H}_5, 76\%$
 $\text{R} = 4\text{-F}, \text{Ar} = \text{C}_6\text{H}_5, 64\%$
 $\text{R} = 4\text{-NMe}_2, \text{Ar} = \text{C}_6\text{H}_5, 28\%$
 $\text{R} = 5\text{-Me}, \text{Ar} = \text{C}_6\text{H}_5, 47\%$
 $\text{R} = 6\text{-OMe}, \text{Ar} = \text{C}_6\text{H}_5, 70\%$

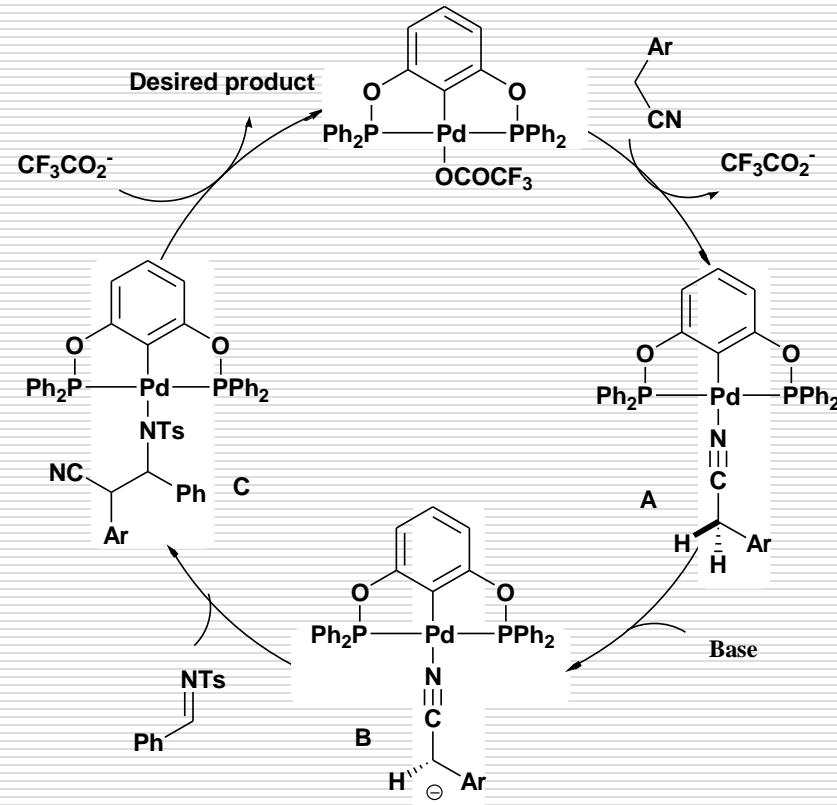


This mechanism is different from that of the rhodium(III)-catalyzed reaction of 2-arylpypyridines with aldimines

4 Addition of C-H bonds to imines

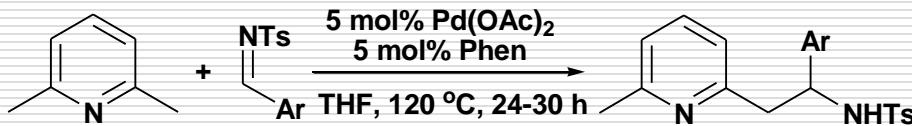


$\text{Ar} = \text{C}_6\text{H}_5$, 99%, dr = 2:1
 $\text{Ar} = 2\text{-FC}_6\text{H}_4$, 95%, dr = 4.6:1
 $\text{Ar} = 2\text{-ClC}_6\text{H}_4$, 99%, dr = 8.4:1
 $\text{Ar} = 2\text{-BrC}_6\text{H}_4$, 83%, dr = 10:1
 $\text{Ar} = 2\text{-IC}_6\text{H}_4$, 99%, dr = 10:1
 $\text{Ar} = 2\text{-MeC}_6\text{H}_4$, 92%, dr = 4:1
 $\text{Ar} = 2\text{-CF}_3\text{C}_6\text{H}_4$, 84%, dr = 4:1
 $\text{Ar} = 2\text{-MeOC}_6\text{H}_4$, 91%, dr = 5.6:1

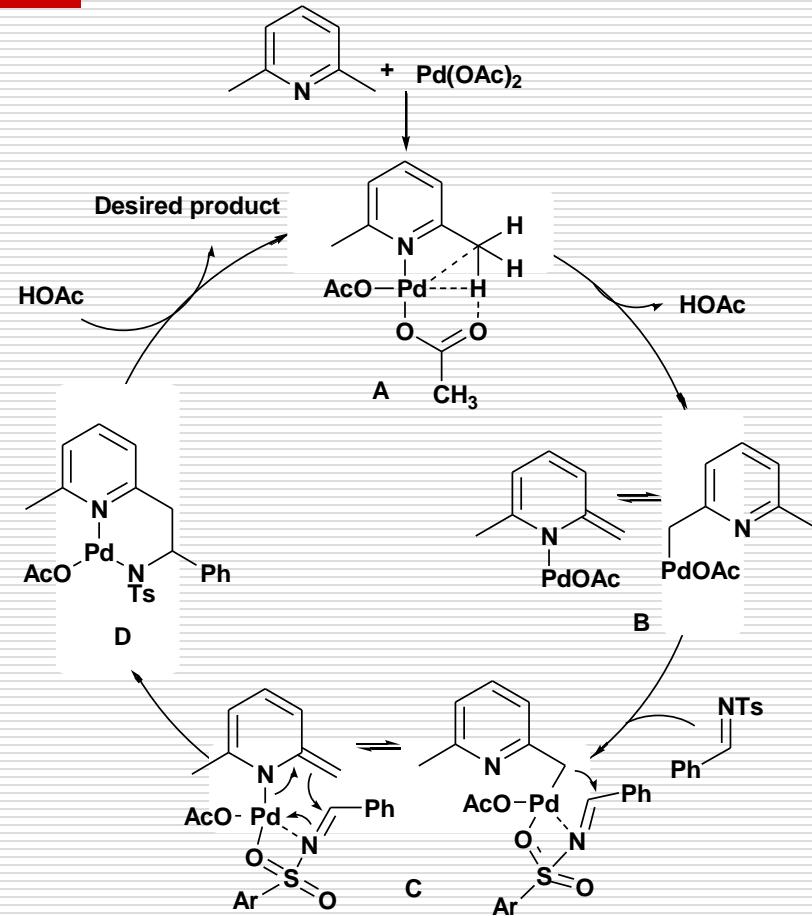


Notes: (1) **mild conditions**
 (2) **high regioselectivity**
 (3) **new palladium-pincer complex**

4 Addition of C-H bonds to imines

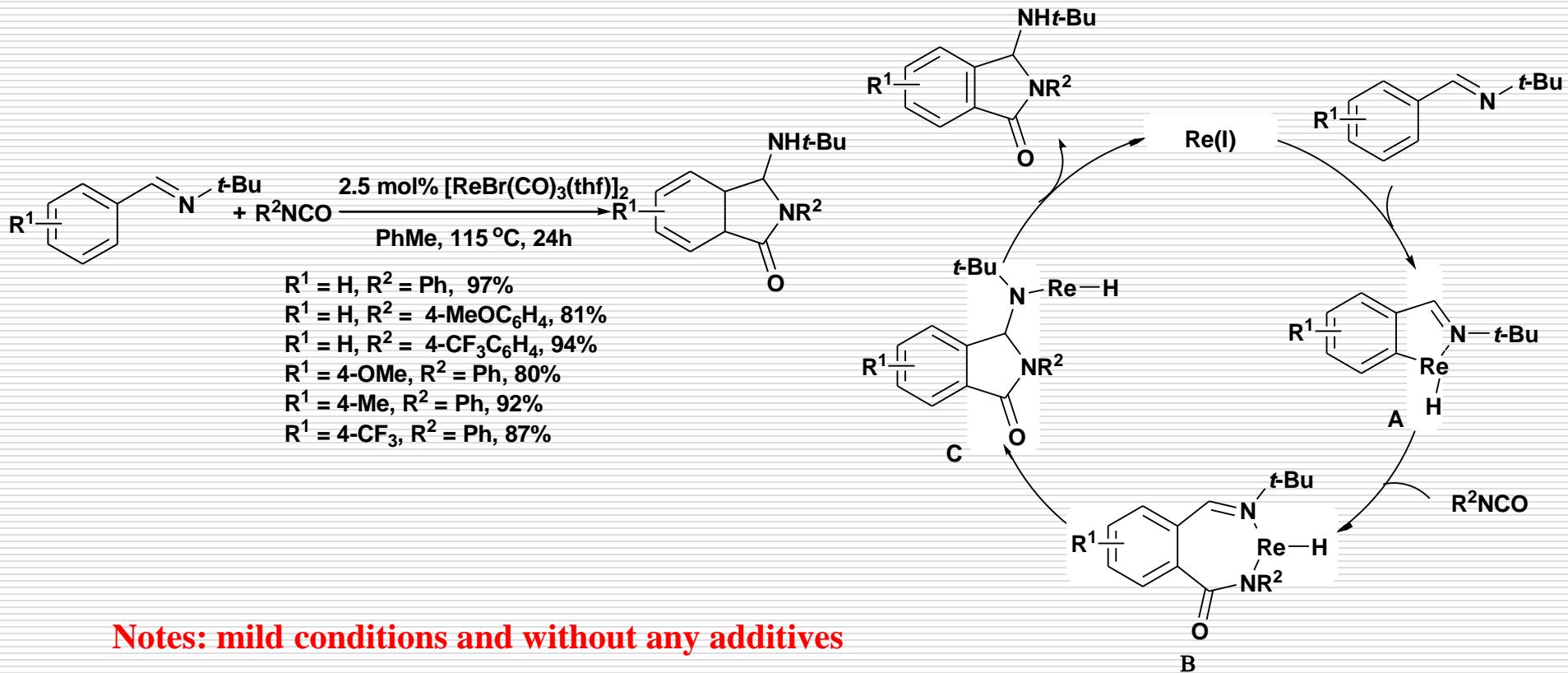


Ar = C_6H_5 , 82%
Ar = 4-ClC₆H₄, 92%
Ar = 4-BrC₆H₄, 85%
Ar = 4-CF₃C₆H₄, 91%
Ar = 4-MeOC₆H₄, 41%
Ar = 2-BrC₆H₄, 86%
Ar = 2-ClC₆H₄, 91%



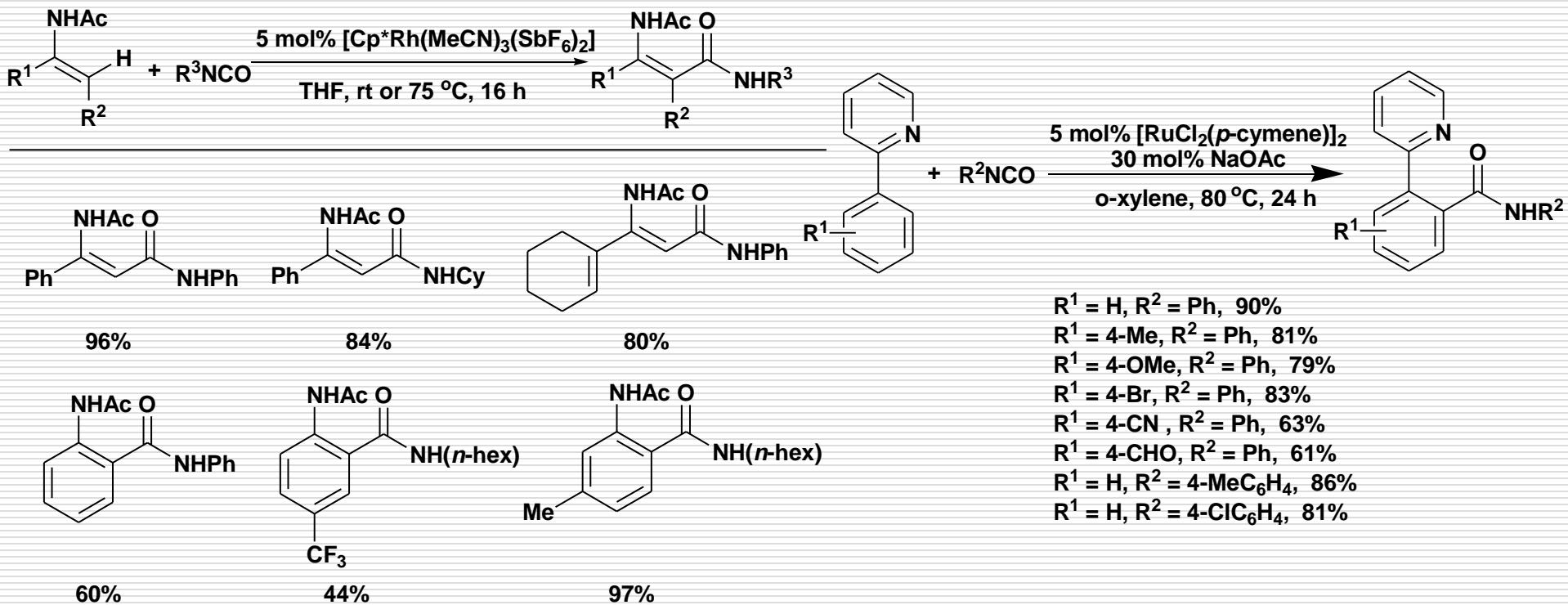
Notes: (1) **mild conditions**
(2) **sp³ C-H bond activation**

5 Addition of C-H bonds to isocyanates

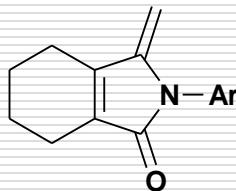
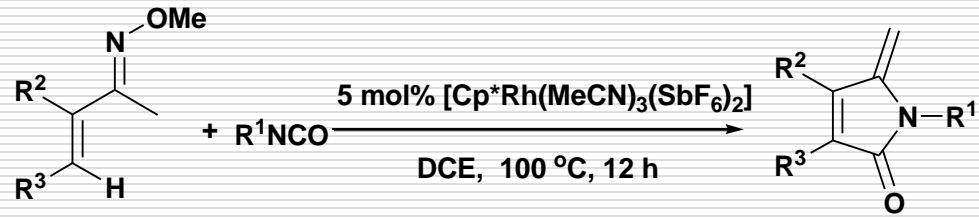


Notes: mild conditions and without any additives

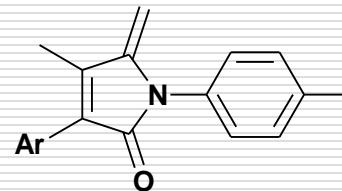
5 Addition of C-H bonds to isocyanates



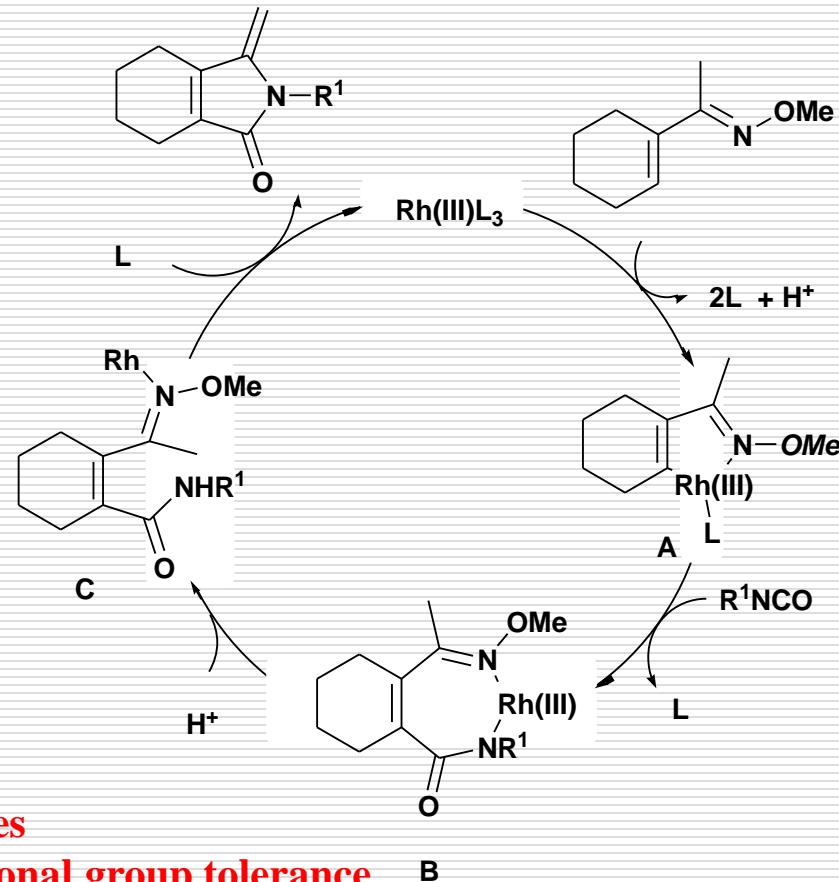
5 Addition of C-H bonds to isocyanates



Ar = 4-MeC ₆ H ₄ , 86%	Ar = 4-CO ₂ EtC ₆ H ₄ , 82%
Ar = 4-MeOC ₆ H ₄ , 82%	Ar = 4-BrC ₆ H ₄ , 92%
Ar = 4-NO ₂ C ₆ H ₄ , 84%	Ar = 4-CIC ₆ H ₄ , 93%
Ar = 4-CF ₃ C ₆ H ₄ , 87%	Ar = 4-FC ₆ H ₄ , 80%



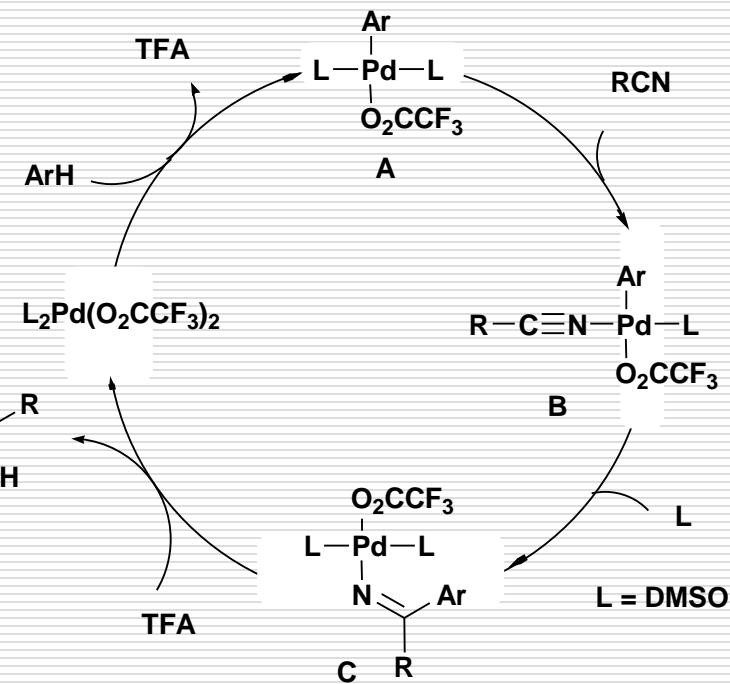
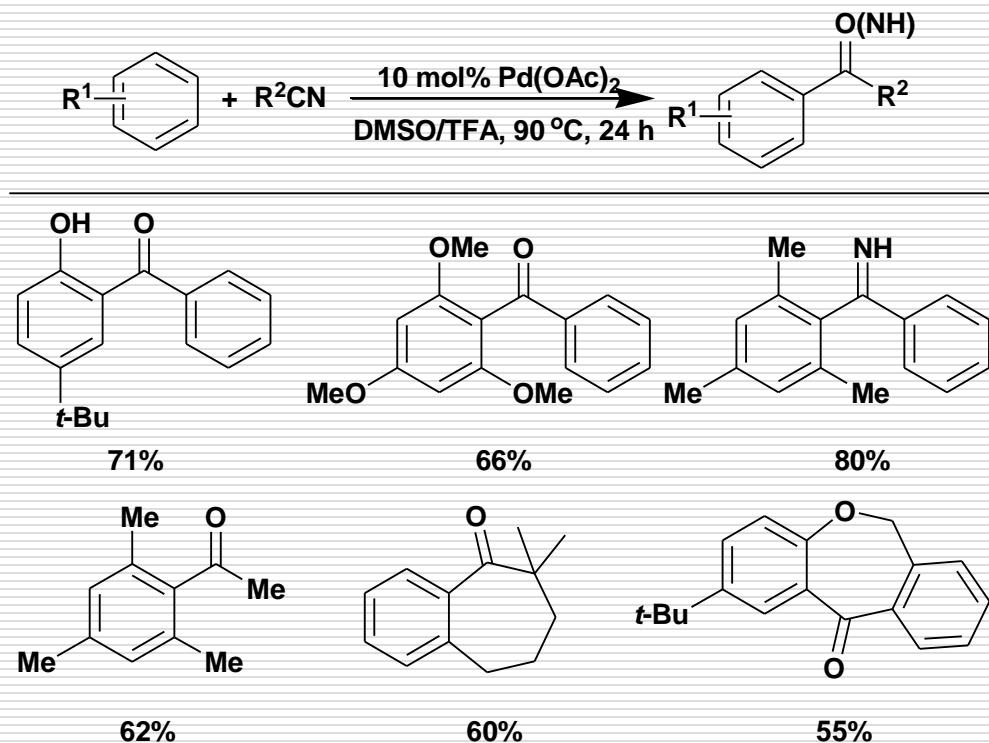
Ar = C ₆ H ₅ , 65%
Ar = 4-MeC ₆ H ₄ , 60%
Ar = 4-MeOC ₆ H ₄ , 57%
Ar = 4-FC ₆ H ₄ , 64%
Ar = 2-Naphthyl, 55%



Notes: (1) **mild conditions and without any additives**

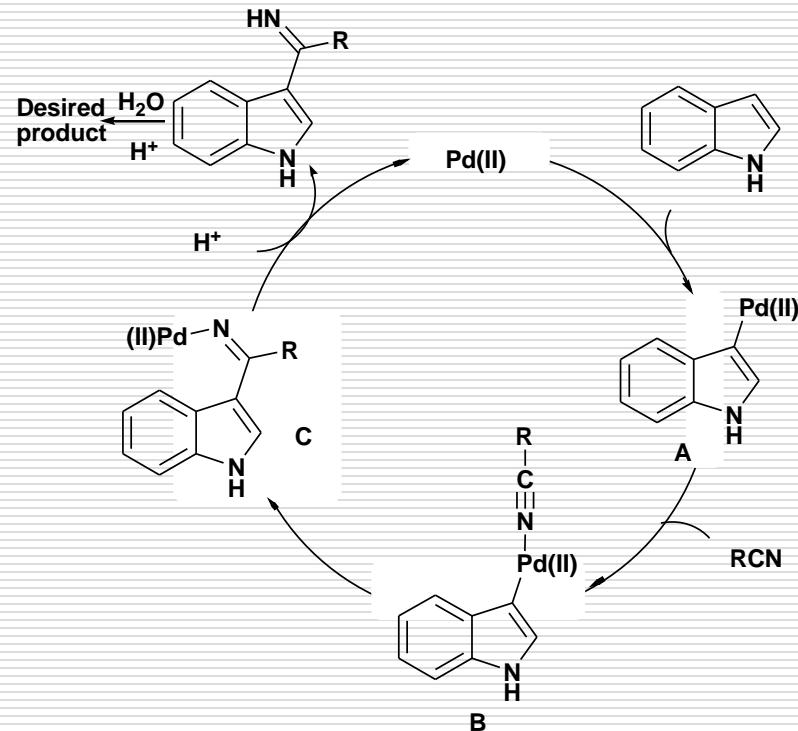
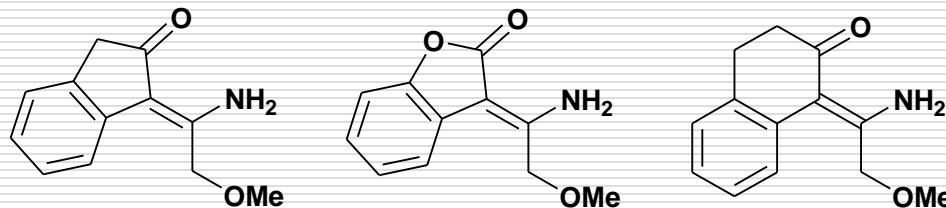
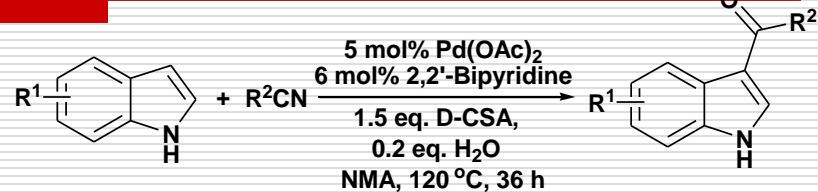
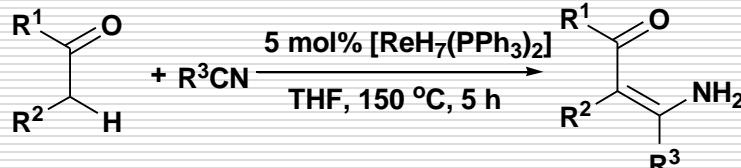
(2) **high regioselectivity and excellent functional group tolerance**

6 Addition of C-H bonds to nitriles



Notes: the reaction can even be improved dramatically by adding only a catalytic amount of DMSO

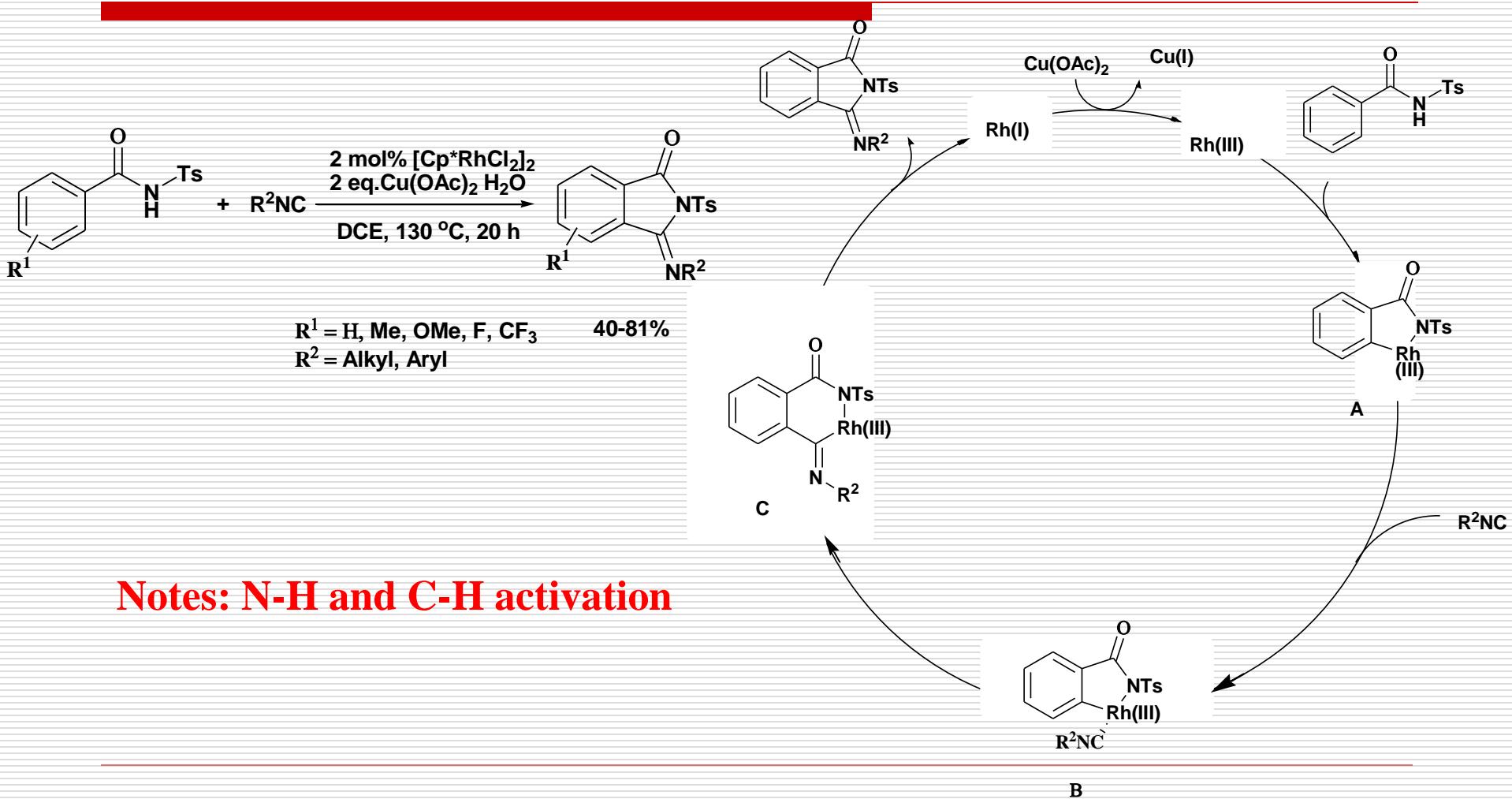
6 Addition of C-H bonds to nitriles



H. Takaya, M. Ito, S.-I. Murahashi, *J. Am. Chem. Soc.* 2009, **131**, 10824-10825.

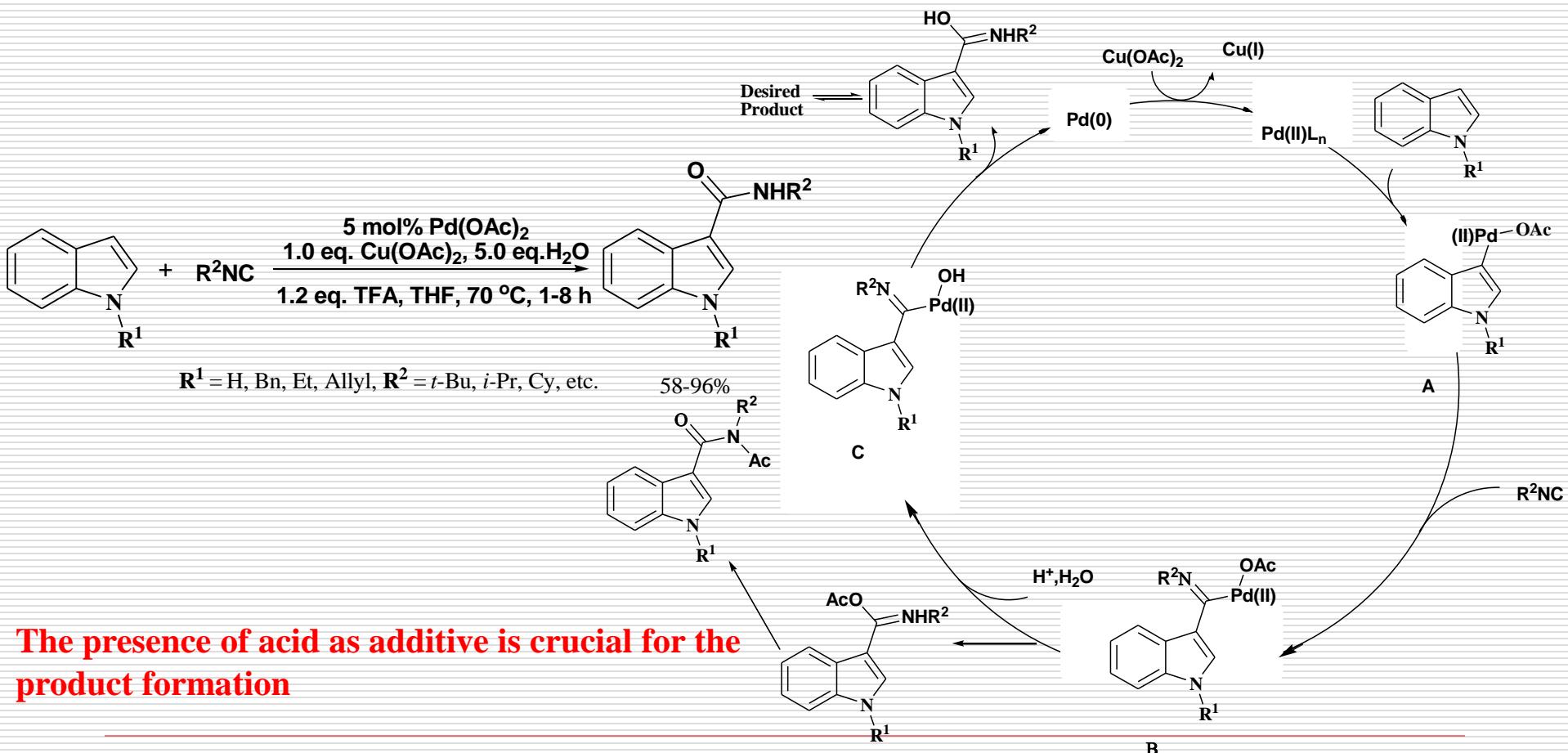
T.-S. Jiang, G.-W. Wang, *Org. Lett.*, 2013, **15**, 788-791

7 Addition of C-H bonds to isocyanides



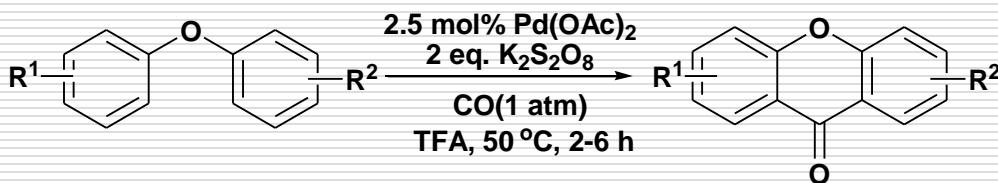
Notes: N-H and C-H activation

7 Addition of C-H bonds to isocyanides

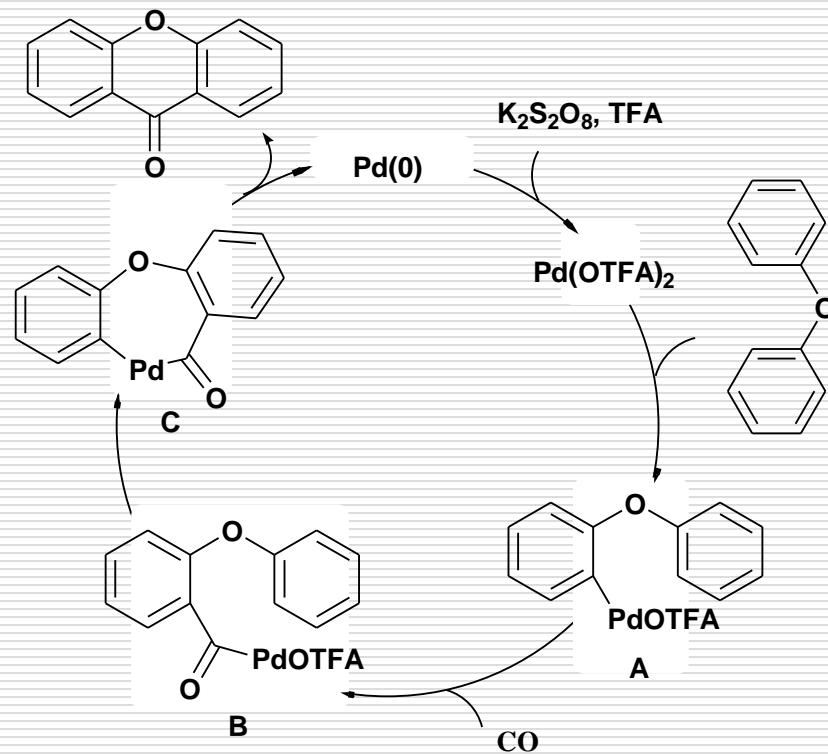


The presence of acid as additive is crucial for the product formation

8 Addition of C-H bonds to carbon monoxide

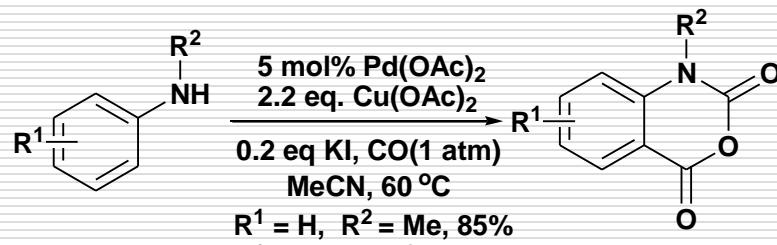


R¹ = 4-Me, R² = 4-Me, 96%
R¹ = 4-Me, R² = 4-t-Bu, 82%
R¹ = 4-Me, R² = 4-CF₃, 56%
R¹ = 4-Me, R² = 4-CO₂E_t, 51%
R¹ = 4-Me, R² = 4-Br, 70%
R¹ = 4-Me, R² = 4-COMe, 58%

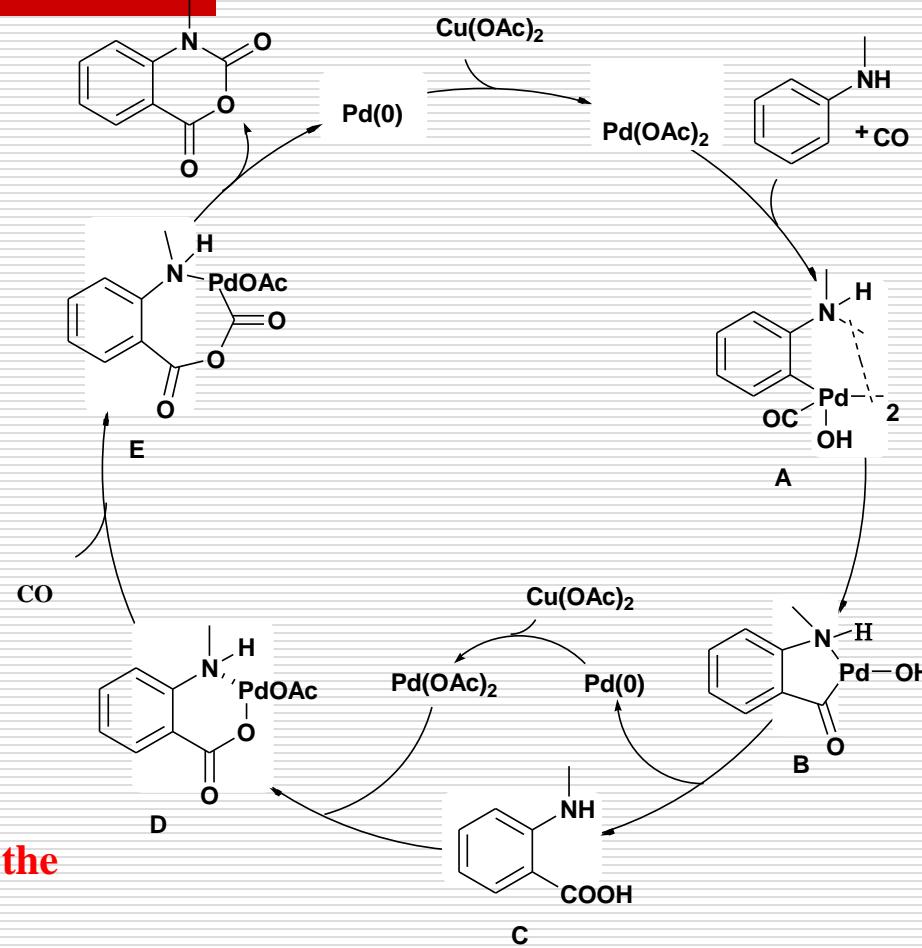


Notes: (1) double C-H functionalization
(2) a simple catalytic system

8 Addition of C-H bonds to carbon monoxide

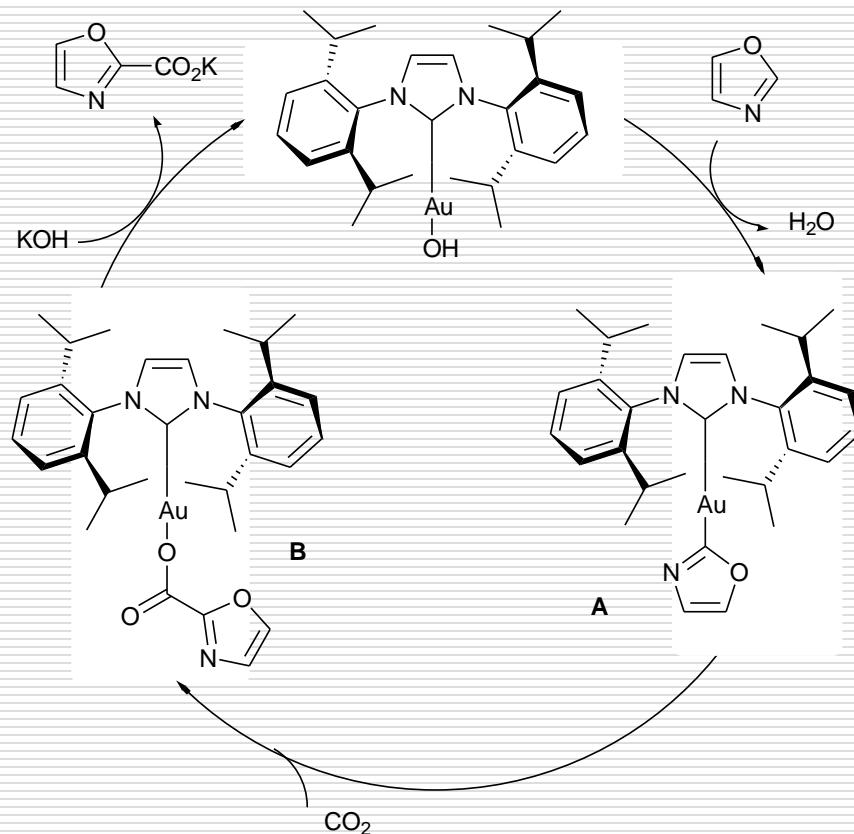
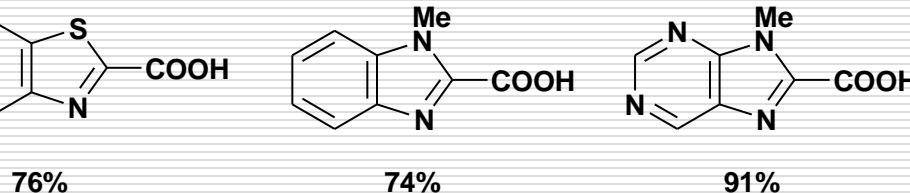
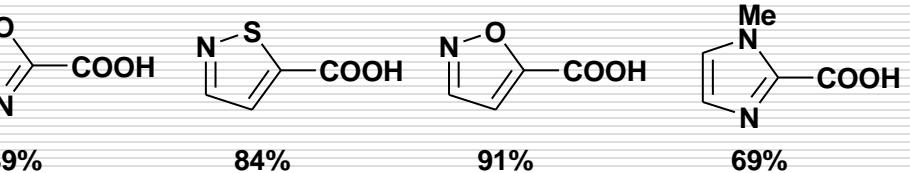
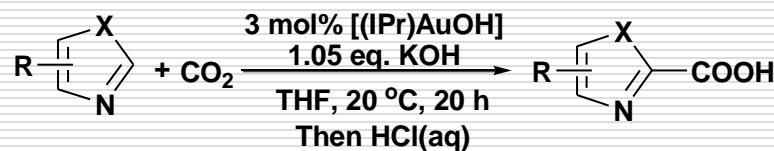


R¹ = H, R² = Me, 85%
R¹ = 4-Me, R² = Me, 76%
R¹ = 4-Br, R² = Me, 86%
R¹ = 3-NO₂, R² = Me, 63%
R¹ = 4-CO₂Et, R² = Me, 85%
R¹ = H, R² = Et, 84%
R¹ = H, R² = Pr, 63%
R¹ = H, R² = Bn, 72%
R¹ = H, R² = Cy, 51%



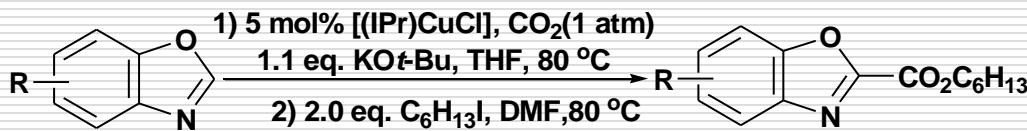
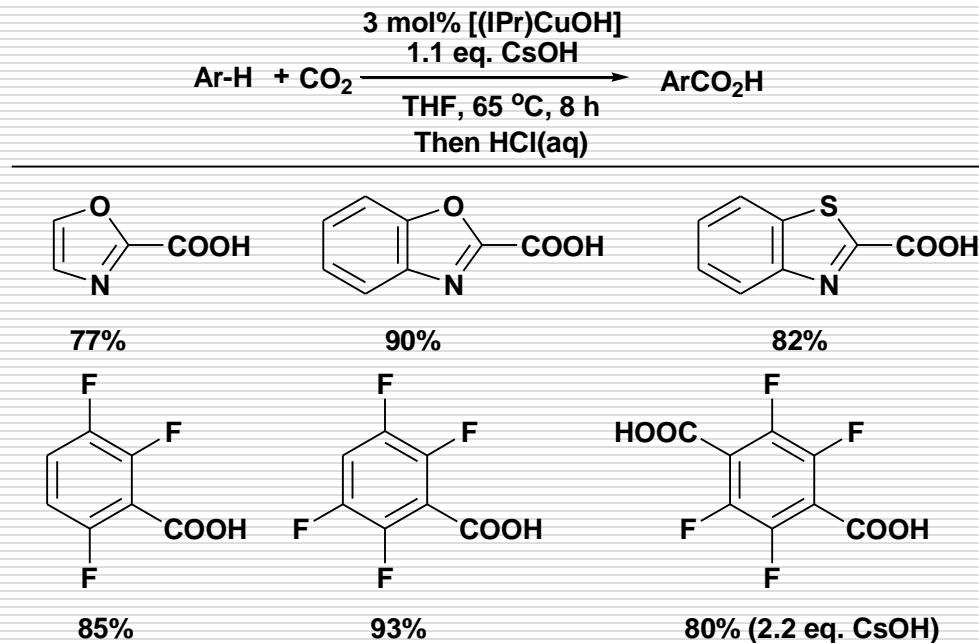
Iodide compounds have been shown to improve the efficiency of Pd-catalyzed carbonylations

9 Addition of C-H bonds to carbon dioxide

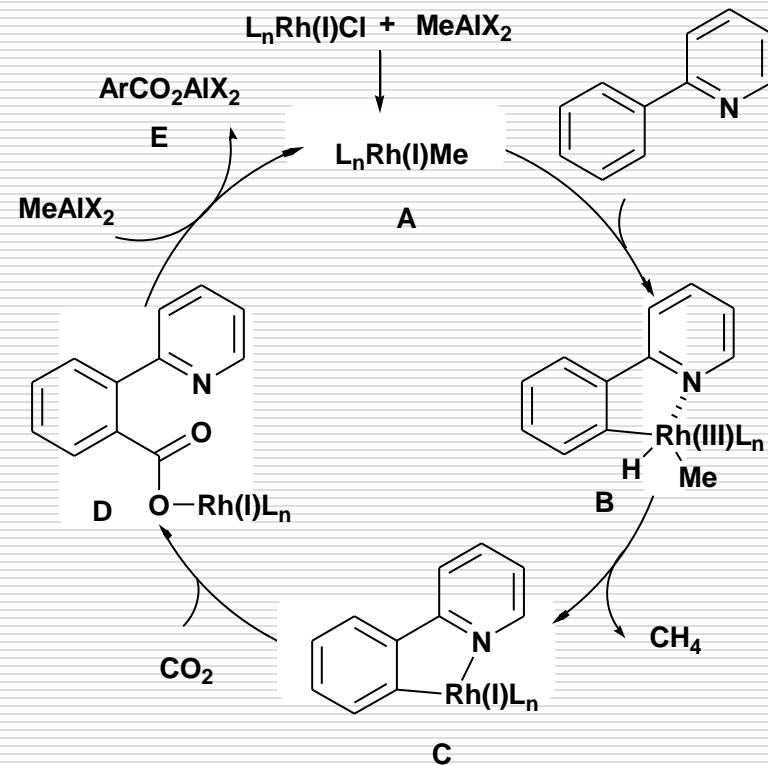
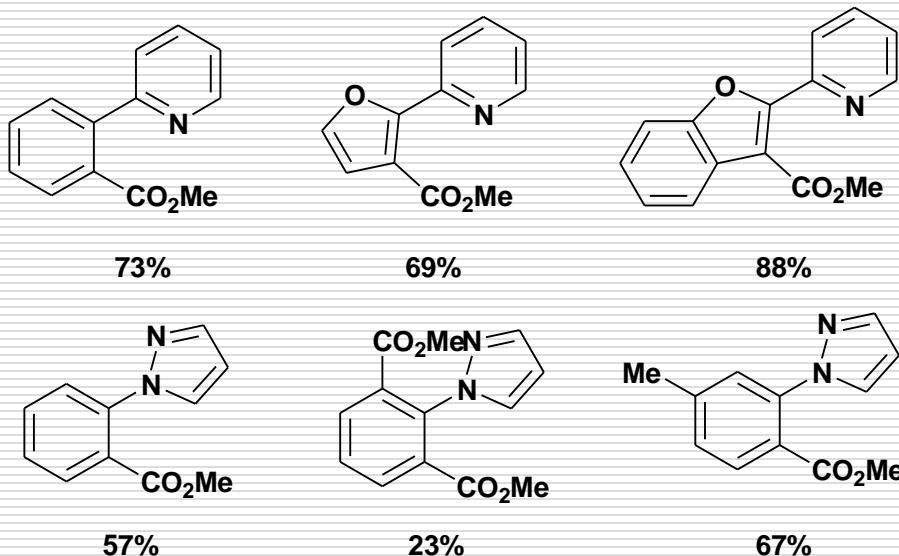
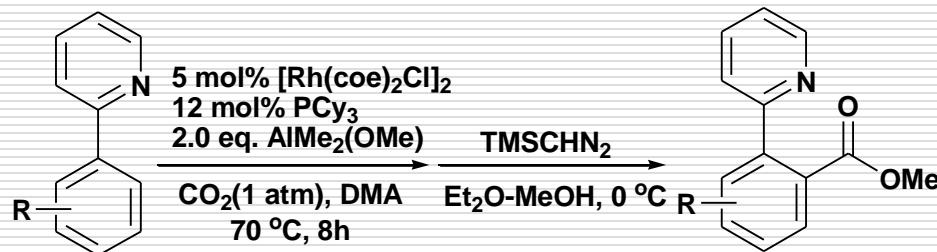


The carboxylation of carbo- and heterocycles with high regioselectivity at the most acidic C-H bond position

9 Addition of C-H bonds to carbon dioxide



9 Addition of C-H bonds to carbon dioxide



10 Conclusions and outlook

- (1) Noble metals: rhodium, ruthenium, rhenium, iridium and palladium.**
 - (2) Most substrates with directing group.**
 - (3) Reactions with chiral ligands could realize the asymmetric catalysis.**
-

Thank you for your attention



Answers of questions

(1)

